Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

## Wan-Yi Liu<sup>a</sup>\* and Yi-Zhi Li<sup>b</sup>

 <sup>a</sup>NinXia Natural Gas Transferring Key Laboratory, NinXia University, Yinchuan
750021, People's Republic of China, and
<sup>b</sup>Coordination Chemistry Institute, State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China

Correspondence e-mail: taixishi@lzu.edu.cn

#### Key indicators

Single-crystal X-ray study T = 273 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.054 wR factor = 0.137 Data-to-parameter ratio = 13.8

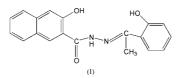
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 2-Hydroxyacetophenone 3-hydroxy-2naphthoylhydrazone

The molecule of the title compound,  $C_{19}H_{15}N_2O_3$ , an aroylhydrazone, is almost planar. The C=O bond length of 1.219 (2) Å suggests that the title compound is in the keto form. The C=N double bond has a length of 1.280 (2) Å. The crystal structure is stabilized by N-H···O, O-H···O and O-H···N hydrogen bonds.

### Comment

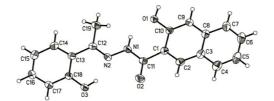
During the past two decades, considerable attention has been paid to the chemistry of aroylhydrazones and their complexes with metal ions (Bu *et al.*, 2001; Liao *et al.*, 2000; Tai *et al.*, 2003); these compounds may serve as potential chelating agents (Fun *et al.*, 1996; Lu *et al.*, 1996) and possess biological activity (Liao *et al.*, 2000). As part of our studies on the synthesis and characterization of aroylhydrazone derivatives, we report the crystal structure of 2-hydroxyacetophenone 3hydroxy-2-naphthoylhydrazone, (I).



In the title compound, (I), the C=O bond length is 1.219 (2) Å, indicating that the molecule is in the keto form. The configuration of the C12–N2 bond is E (Fig. 1). The bond distances and angles in (I) are normal and the molecule is almost planar. As the distances for C12=N2 and C11=O2 are 1.280 (2) and 1.219 (2) Å, respectively, typical for double bonds, this is a novel kind of aroylhydrazone. In the crystal structure, the molecules are stabilized by N–H···O, O–H···O and O–H···N hydrogen bonds (Table 1 and Fig. 2).

#### **Experimental**

All commercially available reagents were used as supplied. The title compound was prepared as follows: a solution of 2-hydroxyaceto-phenone (10 mmol) in ethanol (5 ml) was added to a solution of



#### Figure 1

 $\odot$  2004 International Union of Crystallography Printed in Great Britain – all rights reserved

View of (I), with the atom-labelling scheme and 30% probability displacement ellipsoids.

Received 19 March 2004 Accepted 25 March 2004 Online 31 March 2004 2-hydroxy-3-naphthoylhydrazine (10 mmol) in ethanol (30 ml). The reaction mixture was refluxed for 3 h with stirring, then the resulting pale-yellow precipitate was collected by filtration, washed several times with ethanol and dried *in vacuo* (yield 82%). Analysis calculated for  $C_{19}H_{16}N_2O_3$ : C 71.17, H 4.99, N 8.74%; found: C 71.20, H 4.80, N 8.66%; IR (KBr, cm<sup>-1</sup>): 1661 (C=O), 1642 (C=N). A methanol solution of the title compound was slowly evaporated and pale-yellow crystals were obtained after three weeks.

 $D_x = 1.372 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation Cell parameters from 1742

2226 reflections with  $I > 2\sigma(I)$ 

reflections  $\theta = 2.4-25.5^{\circ}$   $\mu = 0.09 \text{ mm}^{-1}$  T = 273 (2) KBlock, light yellow  $0.25 \times 0.20 \times 0.20 \text{ mm}$ 

 $\begin{aligned} R_{\rm int} &= 0.026\\ \theta_{\rm max} &= 26.0^\circ\\ h &= -10 \rightarrow 7 \end{aligned}$ 

 $k=-9\rightarrow 9$ 

 $l = -27 \rightarrow 27$ 

#### Crystal data

$C_{19}H_{16}N_2O_3$
$M_r = 320.34$
Monoclinic, $P2_1/n$
a = 8.6329 (12)  Å
b = 8.0641 (11)  Å
c = 22.621 (3) Å
$\beta = 99.920(3)^{\circ}$
V = 1551.2 (4) Å <sup>3</sup>
Z = 4

#### Data collection

Bruker SMART APEX CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: none 8027 measured reflections 3039 independent reflections

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0635P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.054$	+ 0.3335P]
$wR(F^2) = 0.137$	where $P = (F_o^2 + 2F_c^2)/3$
S = 1.02	$(\Delta/\sigma)_{\rm max} = 0.038$
3039 reflections	$\Delta \rho_{\rm max} = 0.22 \text{ e} \text{ Å}^{-3}$
220 parameters	$\Delta \rho_{\rm min} = -0.18 \text{ e} \text{ Å}^{-3}$
H-atom parameters constrained	

#### Table 1

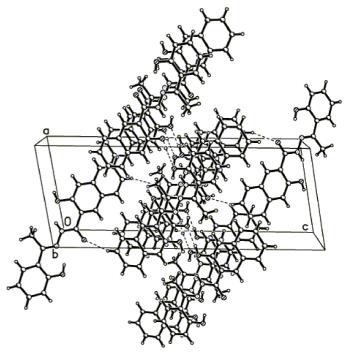
Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N1-H1A···O1	0.86	1.90	2.6026 (18)	137
$O1-H1\cdots O3^i$	0.82	1.89	2.6931 (17)	166
$O3-H3\cdots N2$	0.82	1.77	2.4936 (18)	146

Symmetry code: (i) 1 + x, y, z.

H atoms were positioned geometrically and treated as riding, with C-H = 0.96 (methyl atoms) and 0.93 Å (other atoms), O-H = 0.82 Å, and N-H = 0.86 Å, and with  $U_{\rm iso}(\rm H) = 1.2U_{eq}(\rm C,N)$  and  $1.5U_{\rm iso}(\rm C_{methyl},O)$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve





A view of the crystal packing. Hydrogen bonds are indicated by dashed lines.

structure: *SHELXTL* (Bruker, 2000); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

The authors thank the Natural Science Foundation of Ningxia (02B024) and a State Key Project of Basic Research of the Ministry of Science and Technology (2002).

#### References

- Bruker (2000). *SMART, SAINT* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bu, X. H., Gao, Y. X., Chen, W., Liu, H. & Zhang, R. H. (2001). J. Rare Earths, 19, 70–73.
- Fun, H.-K., Sivakumar, K., Lu, Z.-L., Duan, C.-Y., Tian, Y.-P. & You, X.-Z. (1996). Acta Cryst. C52, 1505–1507.
- Liao, Z.-X., Ma, X.-Y., Shi, Z.-X. & Chen, Y.-Z. (2000). Pol. J. Chem. 8, 1191– 1194.
- Lu, Z.-L., Duan, C.-Y., Tian, Y.-P., You, X.-Z., Fun, H.-K. & Sivakumar, K. (1996). Acta Cryst. C52, 1507–1509.
- Tai, X.-S., Yin, X.-H., Tan, M.-Y. & Li, Y.-Z. (2003). Acta Cryst. E59, o681– 0682.